# **Adsorption versus Absorption of Polychlorinated Biphenyls onto Solid-Phase Microextraction Coatings**

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**Absorption-based polymeric solid-phase microextraction (SPME) fibers with poly(dimethylsiloxane) (PDMS) coatings were used to determine the partitioning coefficients of polychlorinated biphenyls (PCBs) between the sorptive fiber coatings and water. Previous models showing very good correlations between octanol**-**water partitioning coefficients (***K***ow) and absorption-based fiber**-**water partitioning coefficients (***K***dv) for low-molecular-weight ana**lytes failed to predict  $K_{\text{dv}}$  values for PCBs. In fact,  $K_{\text{dv}}$ **values for PCBs were 1**-**7 orders of magnitude lower than those predicted by** *K***ow and actually showed a strong negative correlation between** *K***ow and** *K***dv for higher molecular weight analytes (MW** <sup>&</sup>gt;∼**200).** *<sup>K</sup>***dv values obtained using PDMS fibers with 7- and 100-***µ***m coatings** also disagree, demonstrating that  $K_{\text{dv}}$  cannot be used to **describe the partitioning behavior of PCBs between PDMS and water. However, when PCB partitioning coefficients** were calculated on the basis of surface area  $(K_{ds})$ , the  $K_{ds}$ **values obtained using 7- and 100-***µ***m PDMS fibers agreed reasonably well, demonstrating that surface adsorption is the primary mechanism controlling PCB (and likely other higher molecular weight solutes) partitioning from water to SPME sorbents.**

A number of investigators have reported the correlation between octanol-water partitioning coefficients ( $K_{ow}$ ) and solidphase microextraction (SPME) coating-water partitioning coefficients  $(K_{\text{dv}})$ .<sup>1-5</sup> This correlation was used to predict SPME fiber-water partitioning behavior. These correlations have mostly been applied for analytes having low molecular weights, such as benzene (MW = 78), toluene (MW = 92), and xylenes (MW =

106).<sup>1</sup> However, the correlation between  $K_{ow}$  and  $K_{dw}$ , appears to fail as the molecular weights of the analytes increase, and the poor correlations for analytes such as polycyclic aromatic hydrocarbons (PAHs) and polychlorinated biphenyls (PCBs) have generally been ignored.<sup>4,6,7</sup> These results suggest that higher molecular weight analytes may be *ad*sorbed onto the surface of the polymeric fibers rather than *ab*sorbed into the polymeric phase of the SPME fibers. To investigate the relative importance of these two mechanisms, we determined PCB partitioning coefficients between poly(dimethylsiloxane) (PDMS) (both 7- and 100 *µ*m coatings) and water. Partitioning coefficients were calculated on the basis of sorbent volume  $(K_{\text{dv}})$  and surface area  $(K_{\text{ds}})$ .

#### **EXPERIMENTAL SECTION**

A mixture of 21 PCBs ranging from di- to decachlorobipenyls (100 *µ*g/mL each, AccuStandard Inc., New Haven, CT) was diluted 1:10 000 in acetone to prepare the spiking standard. HPLC-grade water (Fisher Scientific, Pittsburgh, PA) was spiked with 10 *µ*L of the standard solution to achieve a concentration of 50 ppt for each individual congener. This concentration was chosen because it is well within the linear calibration range for all of the PCB congeners<sup>7</sup> and because it is at least an order of magnitude below the solubility limit for the least soluble congeners.<sup>8</sup>

Commercially available 7- and 100-*µ*m film thickness PDMS fibers (Supelco, Bellefonte, PA) were used in this study. Based on the coating dimensions reported by the manufacturer, the volumes of the PDMS phases were 0.026 and 0.612 *µ*L, respectively, while the surface areas of the PDMS fibers were 3.90 and 9.42 mm<sup>2</sup> for 7- and 100- $\mu$ m fibers, respectively. New fibers were used to eliminate the eventual carryover resulting from old fibers.<sup>7</sup> All of the fibers were cleaned for 10 min at 300 °C before each SPME extraction.

The PCB-spiked water was loaded into a 2-mL silanized vial<sup>4,6,7</sup> with a Teflon-lined septum cap, and a magnetic stir bar (8-mm

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**Table 1. Comparison of Volume-Based PDMS/Water Partitioning Coefficients (Kdv) for PCBs Obtained Using 7- and 100-***µ***m Fibers**

			$100 \ \mu m$			5 h $K_{\text{dv}}$ ratio.				
	MW <sup>a</sup>	$K_{\rm ow}$ 8,9	$5 h^b$	24 h <sup>b</sup>	$7 \mu m$ , $5 h^b$	$7 \mu m/100 \mu m$				
PCB-8	223	210 000	$11\,800 \pm 1\,100$	$12\,400 + 2\,300$	$32\,300 \pm 4\,800$	3				
<b>PCB-18</b>	258	440 000	$10\,500 \pm 1\,200$	$10\,800 \pm 1\,900$	$32\,400 \pm 4\,400$	3				
<b>PCB-28</b>	258	550 000	$8700 \pm 870$	$7650 \pm 910$	$35\ 100\pm 6\ 900$	4				
<b>PCB-44</b>	292	4 700 000	$7.670 \pm 1.100$	$6290 \pm 1400$	$55600 \pm 10000$	$\overline{7}$				
<b>PCB-52</b>	292	1800 000	$7270 \pm 620$	$7420 \pm 1900$	$46700 \pm 9600$	6				
PCB-66	292	4 700 000	$8530 \pm 920$	$7500 \pm 1600$	$70300 \pm 13900$	8				
<b>PCB-77</b>	292	3 300 000	$8320 \pm 700$	6 740 $\pm$ 1 500	$83600 \pm 12500$	10				
<b>PCB-101</b>	326	7 100 000	$3750 \pm 630$	$3670 \pm 840$	$35900 \pm 7100$	10				
<b>PCB-105</b>	326		$3130 \pm 640$	$2630 \pm 630$	$36700 \pm 3200$	12				
<b>PCB-118</b>	326	2 500 000	$3310 \pm 660$	$3640 \pm 1100$	$36700 \pm 4900$	11				
<b>PCB-126</b>	326		$1840 \pm 300$	$1670 \pm 390$	$33\,200\pm 5\,000$	18				
<b>PCB-128</b>	361	28 000 000	$932 \pm 210$	$755 \pm 120$	$18300 \pm 1500$	20				
<b>PCB-138</b>	361	10 000 000	$2280 \pm 530$	$2350 \pm 350$	31 100 $\pm$ 2 600	14				
<b>PCB-153</b>	361	28 000 000	$2500 \pm 790$	$2620 \pm 660$	$37500 \pm 5100$	15				
<b>PCB-170</b>	395		$823 \pm 280$	$922 \pm 250$	$17100 \pm 3300$	21				
<b>PCB-180</b>	395		$868 \pm 140$	$825 \pm 270$	$16300 \pm 2900$	19				
<b>PCB-187</b>	395		$1540 \pm 470$	$1830 \pm 330$	$24000 \pm 1500$	16				
<b>PCB-195</b>	430		$510 \pm 140$	$563 \pm 29$	$9930 \pm 500$	19				
<b>PCB-201</b>	430		$967 \pm 220$	$1190 \pm 24$	$17000 \pm 1200$	18				
<b>PCB-206</b>	464	1 400 000 000	$299\pm60$	$279 \pm 19$	6 230 $\pm$ 760	21				
<b>PCB-209</b>	499	9 100 000 000	$252\pm58$	$272 \pm 52$	$5\,650 \pm 1100$	22				
<sup>a</sup> Average molecular weight. $\frac{b}{c}$ SPME/water equilibration time.										

length  $\times$  1.5-mm diameter, Fisher Scientific) was placed inside the loaded sample vial for agitation. *Please note that new stir bars were used for each sample to eliminate the stir bar contamination that was encountered in our previous study.*<sup>7</sup> The fiber was withdrawn inside the needle of the holder, and then the vial septum was pierced with the needle. Once the needle had penetrated the septum of the vial, the plunger on the fiber holder was depressed to expose the fiber to the water sample for 5 or 24 h. The water sample was continuously agitated with the stir bar on a stir plate revolving at ∼1000 rpm during the 5- or 24-h extractions. Once the extraction was completed, the fiber was withdrawn back inside the fiber holder, removed from the water sample, and analyzed immediately.

The extracted PCBs in the fiber were analyzed using an electronic pressure-controlled Hewlett-Packard 5890 Series II gas chromatograph (GC) equipped with a  $63$ Ni electron capture detector (ECD) and a split/splitless injection port. The column was a 25-m HP-5 with an internal diameter of 0.32 mm and a stationary phase thickness of 0.17 *µ*m (Hewlett-Packard, Little Falls, PA). Immediately after the SPME extraction, the SPME fiber was inserted into the 300 °C GC injection port and desorbed for 1 min in the splitless mode. The desorbed PCBs were focused at the column inlet by maintaining the oven at 60 °C during the desorption. Once the desorption was completed, the split vent was opened, and the GC run was started in the normal manner. The initial temperature (60 °C) was ramped at 25 °C/min to 130 °C, then at 8 °C/min to 320 °C, and held at 320 °C for 5 min. The detector temperature was maintained at 300 °C. The GC/ECD response was calibrated by solvent injections of the PCB standard into a cool on-column injector prior to SPME analyses.

**Definition of the Terms and Symbols Used in This Study.** *Absorption.* Analytes are absorbed into the polymeric phase on the SPME fibers, and the partitioning is described by the conventional volume-based distribution coefficient:

$$
K_{\rm dv} = (M_{\rm f}/V_{\rm f})/(M_{\rm w}/V_{\rm w})
$$
 (1)

where  $M_f$  is the mass of analyte absorbed by the fiber,  $V_f$  is the volume of the fiber coating  $(0.026$  and  $0.612 \mu L$  for the 7- and  $100$ - $\mu$ m fibers, respectively),  $M_w$  is the mass of analyte remaining in the water after equilibration, and  $V_w$  is the volume of water.

*Adsorption.* Analytes are accumulated on the surface of the SPME fibers. A surface-based sorption partitioning coefficient is defined as

$$
K_{\rm ds} = (M_{\rm f}/S_{\rm f})/(M_{\rm w}/V_{\rm w})
$$
 (2)

where  $M_f$  is the mass of analyte adsorbed by the fiber,  $S_f$  is the surface area of the fiber coating (3.90 and 9.42 mm<sup>2</sup> for the 7and 100- $\mu$ m fibers, respectively),  $M_w$  is the mass of analyte remaining in the water after equilibration, and  $V_w$  is the volume of water. While  $K_{\text{dv}}$  is unitless,  $K_{\text{ds}}$  has units of centimeters (when the surface area of the fiber is expressed in square centimeters and the water volume is in cubic centimeters).

#### **RESULTS AND DISCUSSION**

PDMS fibers with coating thicknesses of 7 and 100 *µ*m were used to determine the volume-based distribution coefficient  $K_{\text{dw}}$ of PCBs between PDMS and water. Since the 100-*µ*m fiber may require a longer time to reach equilibrium than the 7-*µ*m fiber, the SPME extraction using the 100-*µ*m fiber was performed for 5 and 24 h to ensure equilibration. As shown in Table 1, the  $K_{\text{dv}}$ values obtained from 5- and 24-h SPME extractions agree very well, demonstrating that equilibration was reached within 5 h.

**Table 2. Comparison of Volume-Based PDMS/Water Partitioning Coefficients (Kdv) Obtained Using 7-, 15-, and 100-***µ***m PDMS Fibers, Reported in Refs 4 and 6**

				$K_{\rm dv}$		
	MW	$K_{\alpha w}^4$	100 $\mu$ m <sup>6</sup>	$\mu$ m <sup>6</sup>	15 $\mu$ m <sup>4</sup>	
naphthalene	128	1 0 2 0	705	514	1 0 2 0	
anthracene	178	34 700	1 3 8 0	9 3 8 0	12 600	
benz[a]anthracene	228	407 000	6 7 1 0	28 900	91 200	
benzo[a]pyrene	252	2 750 000	2980	18 400	72 400	
<b>PCB-18</b>	258	437 000			87 100	
<b>PCB-87</b>	326	7 080 000			77 600	

Therefore, 5-h SPME sampling times were used for determining the partitioning coefficients for the remainder of the study.

Unlike previous reports of good agreement between *K*ow and *K*dv for low-molecular-weight analytes (such as benzene, toluene, and xylenes),<sup>1</sup>  $K_{\text{dv}}$  does not agree with  $K_{\text{ow}}$  for PCBs. As shown in Table 1, the  $K_{\text{dv}}$  values decrease with increasing molecular weight of the PCBs, while *K*ow increases dramatically with PCB molecular weight. The  $K_{\text{dv}}$  values range from 1 to 2 orders of magnitude lower than the *K*ow values for the low-molecular-weight PCB congeners to 7 orders of magnitude lower for the highmolecular-weight PCBs.

Negative correlations between  $K_{ow}$  and  $K_{dv}$  for analytes with higher molecular weights are also found in the literature, even though this phenomenon was generally ignored in those reports.<sup>4,6</sup> For example, naphthalene's (MW = 128)  $K_{ow}$  of 1020 is nearly identical to the values of 705, 514, and 1020 (for 100-, 7-, and 15-  $\mu$ m fibers, respectively). However, the agreement becomes much worse with higher molecular weight PAHs. For example, *K*ow and  $K_{\text{dv}}$  for benzo[*a*]pyrene (MW = 252) vary as much as 1000-fold, as shown in Table 2.6 Although only two PCB congeners (PCB-18 and PCB-87) have been reported,<sup>4</sup> the  $K_{\text{dv}}$  values determined by a 15- $\mu$ m PDMS fiber are 5-100 times lower than the  $K_{ow}$  values for the same congener (Table 2), again with the higher molecular weight compound showing the poorest agreement. Based on the  $K_{ow}$  and  $K_{dw}$  data reported in this study and in the literature, we can clearly conclude that  $K_{ow}$  cannot be used to anticipate the  $K_{dw}$ trend in SPME for PCBs and PAHs with molecular weights higher than that of naphthalene.

Comparison of the  $K_{\text{dv}}$  values for PCBs obtained using 7- and 100-*µ*m fibers is also shown in Table 1. Based on the definition of *K*dv, values should not change for the same analyte and the same fiber coating material, regardless of the coating thickness. However, Table 1 shows that the  $K_{\text{dv}}$  values of PCBs in a PDMSwater system obtained using a 7-µm fiber are 3–22 times higher than those obtained using the  $100 \mu m$  fiber. The disagreement in  $K_{\text{dv}}$  values obtained using the same PDMS fibers was also reported in the literature for high-molecular-weight PAHs.<sup>6</sup> As shown in Table 2, the  $K_{\text{dv}}$  values obtained using 7- and 15- $\mu$ m fibers are 4-24 times higher than those obtained using the 100-*µ*m fiber for higher molecular weight PAHs (e.g., anthracene,  $MW = 178$ ;  $benz[a]$ anthracene, MW = 228; and  $benzo[a]$ pyrene, MW = 252), although the  $K_{\text{dv}}$  values obtained using all three fibers agree reasonably well for naphthalene ( $MW = 128$ ). The disagreement in  $K_{\text{dv}}$  values for PDMS fibers with different coating thicknesses clearly demonstrates that  $K_{\text{dv}}$  is not valid to describe the sorption behavior of analytes having higher molecular weights.

#### **Table 3. Comparison of Surface-Based Fiber/Water Partitioning Coefficients (Kds) for PCBs Obtained Using PDMS-Coated Fibers**



<sup>a</sup> The surface areas are 3.90 and 9.42 mm<sup>2</sup> for 7- and  $100 \text{-} \mu \text{m PDMS}$ fibers, respectively.  $K_{ds}$  is defined in eq 2, where the surface area of the fiber is expressed in  $\text{cm}^2$  and the water volume is in  $\text{cm}^3$ . *b*  $K_{ds}$ ratio is the  $K_{ds}$  value determined for the  $7\mu$ m fiber divided by the  $K_{ds}$ value determined for the 100-*µ*m PDMS fiber.

#### **Table 4. Comparison of Surface-Based PDMS/Water Partitioning Coefficients (Kds) for PAHs Obtained Using 7- and 100-***µ***m PDMS Fibers**



 $a K<sub>ds</sub>$  values were calculated using the  $K<sub>dv</sub>$  values in Table 2 from ref 6 and the surface areas of the fibers.

Since the difference between the  $K_{\text{dv}}$  values for the two PDMS fibers increases with increasing molecular weight of the PCBs, the sorption mechanism of PCBs in a PDMS-water system may be controlled more by sorbent surface area than by sorbent volume. Therefore, we calculated the surface partitioning coefficient,  $K_{ds}$ , based on the fiber surface area (eq 2). As shown in Table 3, except for the lowest molecular weight PCBs, the  $K_{ds}$ values obtained using the 7- and 100-*µ*m PDMS fibers agree much better than the  $K_{\text{dv}}$  values (Table 1). We also calculated the  $K_{\text{ds}}$ values on the basis of the  $K_{\text{dv}}$  values of PAHs obtained using 7and 100-*µ*m PDMS fibers and their surface areas.6 As shown in Table 4, the  $K_{ds}$  values of the 7- and 100- $\mu$ m fibers agree well for the higher molecular weight PAHs, while the  $K_{\text{dv}}$  values determined using the 7- and 100-*µ*m fibers do not agree well (Table 2). The better agreement in  $K_{ds}$  than in  $K_{dv}$  values between the 7- and 100- $\mu$ m PDMS fibers clearly demonstrates that  $K_{ds}$  is more valid than  $K_{\text{dv}}$  for describing the sorption mechanisms of PCBs and PAHs in the SPME process. Thus, the dominant sorption

process for PCBs and PAHs on PDMS coatings is adsorption, especially for the higher molecular weight species.

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